**Periodic Table Template**

Follow the instructions from your teacher for filling in the electrons in the shells of these elements. Write in the electron arrangement below the diagram. Aluminium is done for you as a guide.

|  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- |
| 1  H |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  | 2 He |
| 3Li | 4Be |  |  |  |  |  |  |  |  |  |  | 5 B | 6 C | 7 N | 8 O | 9 F | 10Ne |
| 11Na | 12Mg |  |  |  |  |  |  |  |  |  |  | 13 Al 2, 8, 3 | 14 Si | 15 P | 16 S | 17 Cl | 18Ar |
| 19 K | 20 Ca | 21 Sc | 22 Ti | 23 Va | 24 Cr | 25Mn | 26 Fe | 27 Co | 28 Ni | 29 Cu | 30 Zn | 31 Ga | 32 Ge | 33 As | 34 Se | 35 Br | 36 Kr |

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| --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- |
| 1  H |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  | 2 He |
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